

A Convenient Preparation of
2-Fluoro-3-alkoxy-1,3-butadienes

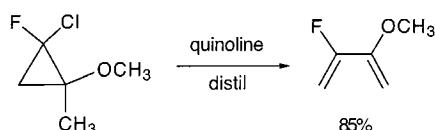
Timothy B. Patrick,* James Rogers, and Keith Gorrell

Department of Chemistry, Southern Illinois University, Edwardsville, Illinois 62026

tpatric@siue.edu

Received July 13, 2002

ABSTRACT



1-Chloro-1-fluoro-2-methoxy-2-methylcyclopropane eliminates HCl on heating in quinoline solution above 50 °C to give 2-fluoro-3-methoxy-1,3-butadiene in high yield. If an alcohol is added to the reaction then a 2-fluoro-3-alkoxy-1,3-butadiene is obtained in high yield. The dienes give smooth 4 + 2 cycloaddition reactions.

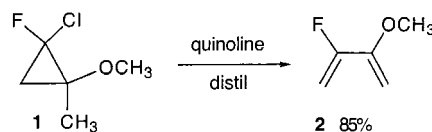
Methodology for the preparation of monofluoro 1,3-butadienes consists of fluorocyclopropane ring opening,^{1–4} palladium(0) crossed coupling reactions of fluorinated alkenyl stannanes,^{5,6} and Wittig olefination of fluorinated α,β -unsaturated aldehydes.⁷

In 1993, Johnson and co-workers proposed a fluorodiene intermediate in the preparation of fluorinated polyenes from reaction of 1-fluoro-1-chloro-2-methoxy-2-methylcyclopropane (**1**)⁸ with a dienol in a solution of toluene and pyridine at 115 °C for 10 h.⁴

We report here that when **1** is heated in pyridine solution a clear colorless distillate of 2-fluoro-3-methoxy-1,3-butadiene (**2**)⁹ is obtained around 50–80 °C in 85% yield.

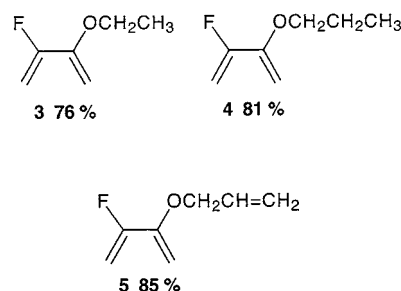
Pyridinium hydrochloride is obtained from the reaction mixture on cooling. The product is obtained with a small contamination with pyridine, but when the solvent is changed to quinoline **2** is obtained pure as seen in Scheme 1.

Scheme 1



When the reaction is performed in the presence of an excess of alcohol (ethanol, propanol, or allyl alcohol), then 2-fluoro-3-ethoxy-1,3-butadiene (**3**), 2-fluoro-3-*n*-propyl-1,3-

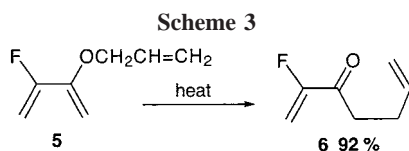
Scheme 2



- (1) Shi, G.-q.; Schlosser, M. *Tetrahedron* **1993**, *49*, 1456.
- (2) Schlosser, M.; Dahan, R.; Cottens, S. *Helv. Chim. Acta* **1994**, *67*, 284.
- (3) Kobayashi, Y.; Morikawa, T.; Youshizawa, A.; Taguchi, T. *Tetrahedron Lett.* **1981**, *22*, 5297.
- (4) Johnson, W. S.; Chenera, B.; Tham, F. S.; Kulling, R. K. *J. Am. Chem. Soc.* **1993**, *115*, 493.
- (5) Chen, C.; Wilcoxon, K.; Kim, K.-i.; McCarthy, J. R. *Tetrahedron Lett.* **1997**, *38*, 7677.
- (6) Lu, L.; Burton, D. J. *Tetrahedron Lett.* **1997**, *38*, 4673.
- (7) Kondo, K.; Cottons, S.; Schlosser, M. *Chem. Lett.* **1984**, *12*, 2149.
- (8) Bessiere, Y.; Savary, D. N.-H.; Schlosser, M. *Helv. Chim. Acta* **1977**, *173*, 1739.
- (9) All new compounds (**2**–**8**) gave spectral data consistent with the assigned structure. Compound **2**: ¹H NMR (CDCl₃, TMS) δ 3.62 (s, CH₃), 4.2–5.0 (m, vinyl); ¹⁹F NMR (TFA=O, 76.6 relative to CFCl₃) δ –40.0; ¹³C NMR δ 57.5 (CH₃), 84.1 (C4), 90.5 (d, C1), 153.8 (d, C3), 158.3 (d, C2, *J* = 189 Hz). Compound **6**: ¹H NMR δ 2.38 (m, 4H, CH₂), 4.9–6.05 (m, 5H, vinyl); ¹⁹F NMR –116.2; ¹³C NMR δ 27.1 (C4), 43.1 (d, C3), 100.5 (C1), 114.9 (C7), 136.5 (C6), 159.5 (d, *J* = 259 Hz), 206 (d, C1). Compound **7**: ¹H NMR δ 4.1 (m, CH₂), 4.21 (m, CH₂), 7.25 (m, aromatic); ¹⁹F NMR –71; Compound **8**: ¹H NMR δ 2.84 (s, OCH₃), 3.63, 4.00 (two multiplets, CH₂ next to F), 3.85 (m, CH₂ next to OCH₃); ¹⁹F NMR –58.6.

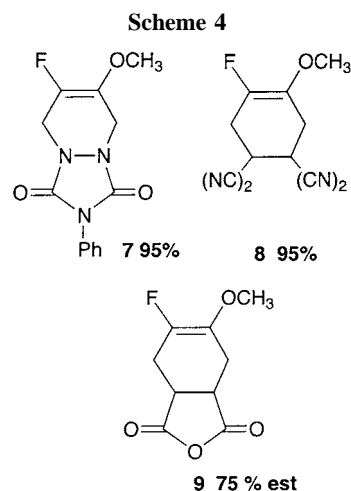
butadiene (**4**), and 2-fluoro-3-allyloxy-1,3-butadiene (**5**) are obtained in good yields. (Scheme 2).

When compound **5** was heated above 80 °C for 30 min, the product from Claisen rearrangement (**6**) was obtained in 92% yield (Scheme 3).



When diene **2** is subjected to reaction with reactive dienophiles such as *N*-phenyl-1,2,4-triazoline-3,5-dione and tetracyanoethylene, the 4 + 2 cycloaddition products **7** and **8** were obtained respectively in good yield (Scheme 4). Reaction of **2** with maleic anhydride furnished the cycloadduct **9** in a mixture of products. These reactions were performed in CDCl₃ solvent at room temperature over 6 h.

Acknowledgment. We thank the National Science Foundation (RUI) for support of this research.



Supporting Information Available: ¹H NMR and analytical data for **2–8**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL026512V